REMARKS

Claims 93-110 are pending in the present application. Applicants thank the Office for indicating that the subject matter of Claims 97 and 98 is allowable if rewritten in independent form.

The Office rejected Claims 93-96 and 99-110 over Worrel (U.S. 3,948,619) in combination with Cherpeck (U.S. 5,300,701) and Baxter (U.S. 6,562,913). The Office cites to Worrel as evidence that is would be obvious to use a polyisobutene molecule to form a Mannich adduct.

Present independent Claim 93 recites a step of "alkylating a phenol with a highly reactive polyisobutene ...". The alkylating of the presently claimed invention is carried out on a polyisobutene, not a polybutene. At best, <u>Worrel</u> describes an "olefin reactant" that is made by polymerizing butene (see column 5, lines 24-36 of <u>Worrel</u>). Nowhere in the <u>Worrel</u> patent is a polyisobutene material described. In fact, the Office even admits that <u>Worrel</u> "differs from the claims in that he does not specifically teach that the alkyl group is a highly reactive PIB ..." [polyisobutene] (see the last paragraph on page 3 of the August 19 Office Action).

The Office attempts to remedy this defect of <u>Worrel</u> by citing to <u>Cherpeck</u> and/or <u>Baxter</u>. Applicants submit that <u>Cherpeck</u> is likewise deficient with respect to the Office's assertion of obviousness. Like <u>Worrel</u>, <u>Cherpeck</u> does not disclose or suggest all of the present claim limitations. At best <u>Cherpeck</u> discloses a polyisobutene substituted phenolic compound, not a Mannich adduct thereof. This disclosure, however, is insufficient to render the presently-claimed invention obvious because the olefin reactant of <u>Cherpeck</u> is not the Mannich adduct recited in the present claims.

Baxter may disclose or suggest a process for making a highly reactive polyisobutene but is silent with respect to any process for using the thus-obtained polyisobutene to alkylate a phenol and use the thus-alkylated phenol to form a Mannich adduct.

While the art of record, e.g., <u>Cherpeck</u> and/or <u>Baxter</u>, may disclose that polyisobutenes were known to those of skill in the art at the time the present application was filed, the cited art fails to provide any motivation for one of skill in the art to modify <u>Worrel</u> to arrive at the presently claimed invention. First, the cited art provides no reason why one of skill in the art would beieve that a polyisobutene such as that disclosed or suggested in any of <u>Cherpeck</u> and/or <u>Baxter</u> would in fact be an equivalent substitute for any polybutene disclosed in <u>Worrel</u>. A polyisobutene has a structure that is substantially different than a generic polybutene. Polyisobutene includes two methyl (-CH₃) groups which are pendent to a polymer chain. In contrast, a polybutene contains a single pendent ethyl (-CH₂CH₃) substituent. Nowhere in the art of record is there any evidence proving that a polyisobutene and a polybutene are readily exchangeable and/or have equivalent physical and/or chemical properties.

Applicants submit that the Office's assertion that it would be obvious to use any polyisobutene disclosed or suggested in <u>Cherpeck</u> and/or <u>Baxter</u> in the process of <u>Worrel</u> or as a substitute for the <u>Worrel</u> polybutene is not supportable and the rejection should be withdrawn.

The differences in the alkyl substituent of <u>Worrel</u> and that of the present claims extends likewise to the relative molecular weights and amounts of vinylidene bonds present in the polybutene of <u>Worrel</u> in comparison to the polyisobutene of the present claims. It is an express requirement of present Claim 1 that the highly reactive polyisobutene has a vinylidene double bond content of "more than 70 mol%" and a molecular weight of "less than 900". The highly reactive polyisobutene of the present claims also has a polydispersity

requirement. When considered as a whole, <u>Worrel</u> is not suggestive of such a particular polyisobutene. In fact, <u>Worrel</u> discloses that the polybutene most preferred for the <u>Worrel</u> process is one having a molecular weight of 900-1,100 (see column 5, line 34 of <u>Worrel</u>), i.e., not encompassed by the "less than 900" molecular weight of the polyisobutene of the present claims.

The <u>Worrel</u> patent further does not disclose with sufficient specificity the other reactants recited in the present claims. For example, present Claim 93 recites an amine of formula NHR⁴R⁵ where both R⁴ and R⁵ are C₁-C₂₀ alkyl radicals. At best, <u>Worrel</u> discloses generic amines but nowhere discloses that any dialkyl amine (e.g., an amine of formula -NR¹R²) is preferred or provides any advantages. To the contrary, <u>Worrel</u> describes diamines (e.g., an amine having at least two N atoms). The amine described in the present claims is a dialkyl monoamine which is distinct and separate from the diamine of <u>Worrel</u>.

Applicants draw the Office's attention to the examples in co-pending application 10/089,064. The examples in co-pending '064 describe the effect obtained when a dialkyl monoamine (e.g., a secondary monoamine) is used in certain fuel formulations. Pages 35-36 of copending '064 are especially relevant in this regard. Table 2 on page 36 of co-pending '064 describes different fuel additive compositions which contain the monoamine of the present claims or a diamine. For example, Comparative Example 2 includes ethylene diamine (EDA), one of the preferred amines of Worrel. Inventive Example 2 includes dimethylamine, a dialkyl monoamine that conforms to the amine recited in present Claim 93.

The effect of the amine on the performance of the fuel additive is described in the columns identified as "Valve Deposits" in Table 2 on page 36 of co-pending '064. Inventive Example 2 which includes a monoamine that adheres to the formula for the amine of present Claim 93 has the best performance. The ethylene diamine-containing comparative example is a composition which forms substantially greater amounts of valve deposits. Where the

amine value of the valve deposits for the comparative example is 84 mg/valve, the mean

value for the valve deposits formed when a dialkyl monoamine is used is 0 mg/valve.

Applicants submit that the factual evidence in co-pending '064 is probative of the

patentability of the presently claimed subject matter. Contrary to Worrel's disclosure that a

diamine is a preferred amine, Applicants have shown that substantially improved

performance is obtained when a monoamine is used for making a Mannich adduct-containing

composition.

Applicants submit the Office's reliance on Worrel, Cherpeck and Baxter as evidence

that the presently claimed invention is obvious amounts to nothing more than hindsight

reasoning. Applicants submit the presently claimed invention is not obvious for the reasons

discussed above, especially in view of Applicants' factual evidence rebutting the Office's

assertion of obviousness, and respectfully request withdrawal of the rejection.

For the reasons discussed above in detail, Applicants request withdrawal of the

rejection and the allowance of all now-pending claims.

Respectfully submitted,

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